

Soils of Hamakua

Concepts in Soil Fertility

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Outline

- Importance of Soils
- Concepts in Soil Fertility
 - Clays
 - CEC
 - Organic matter
 - pH
 - N and P
- Soils of Hamakua

Medium for
Plant growth



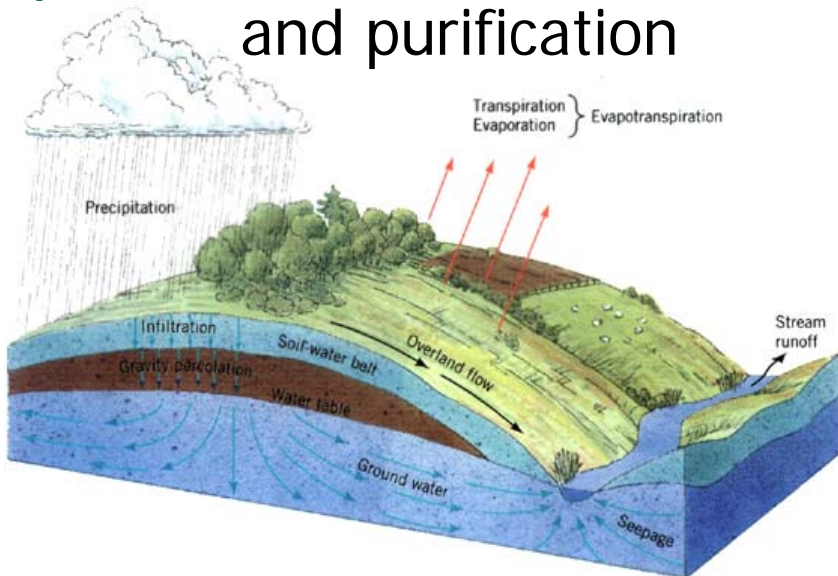
Recycling
system



Habitat for
Soil organisms

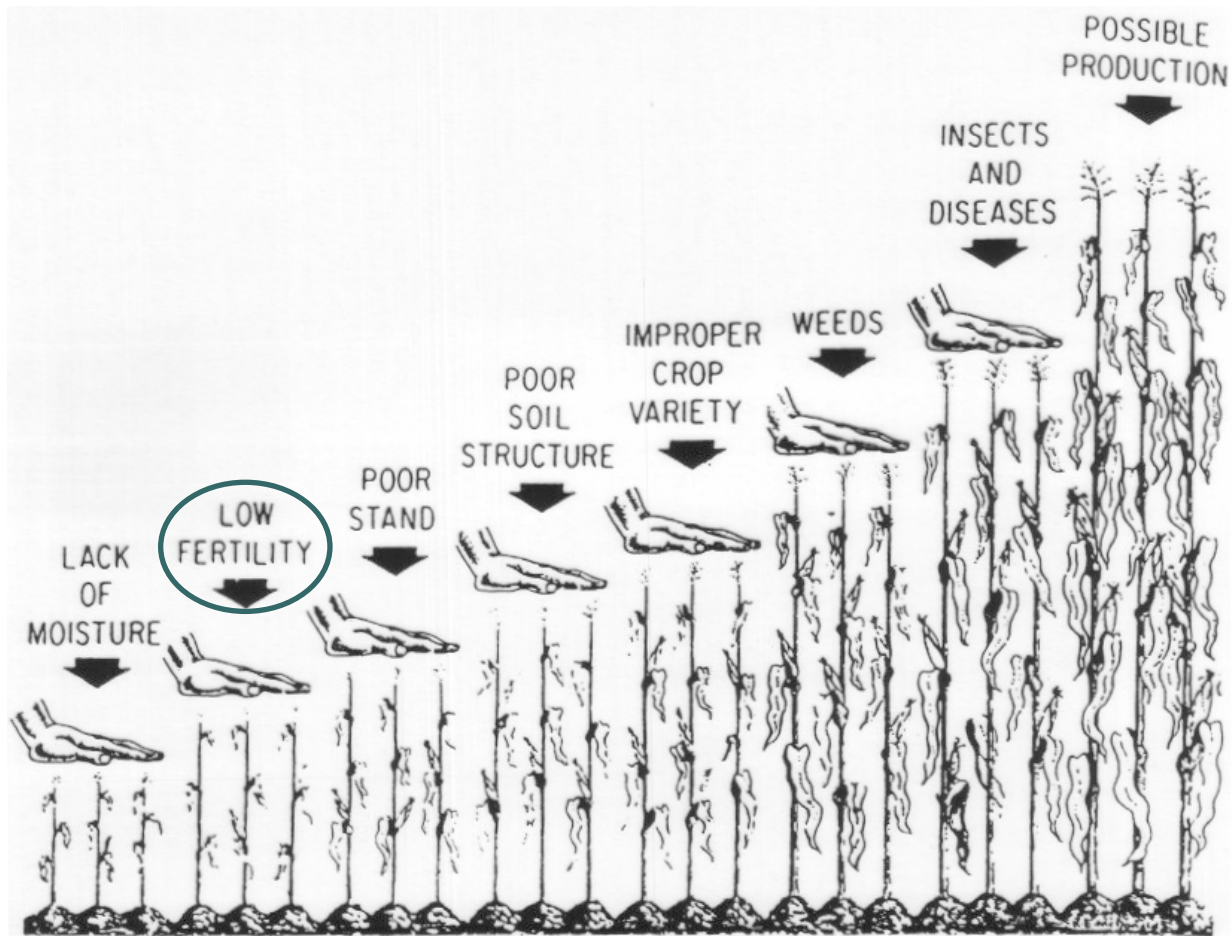
5 Functions
of Soil

Water supply
and purification



Engineering Medium

Plant Growth Factors



Essential Plant Nutrients

Air and water: C, H, O

Macronutrients: N, P, K, Ca, Mg, S

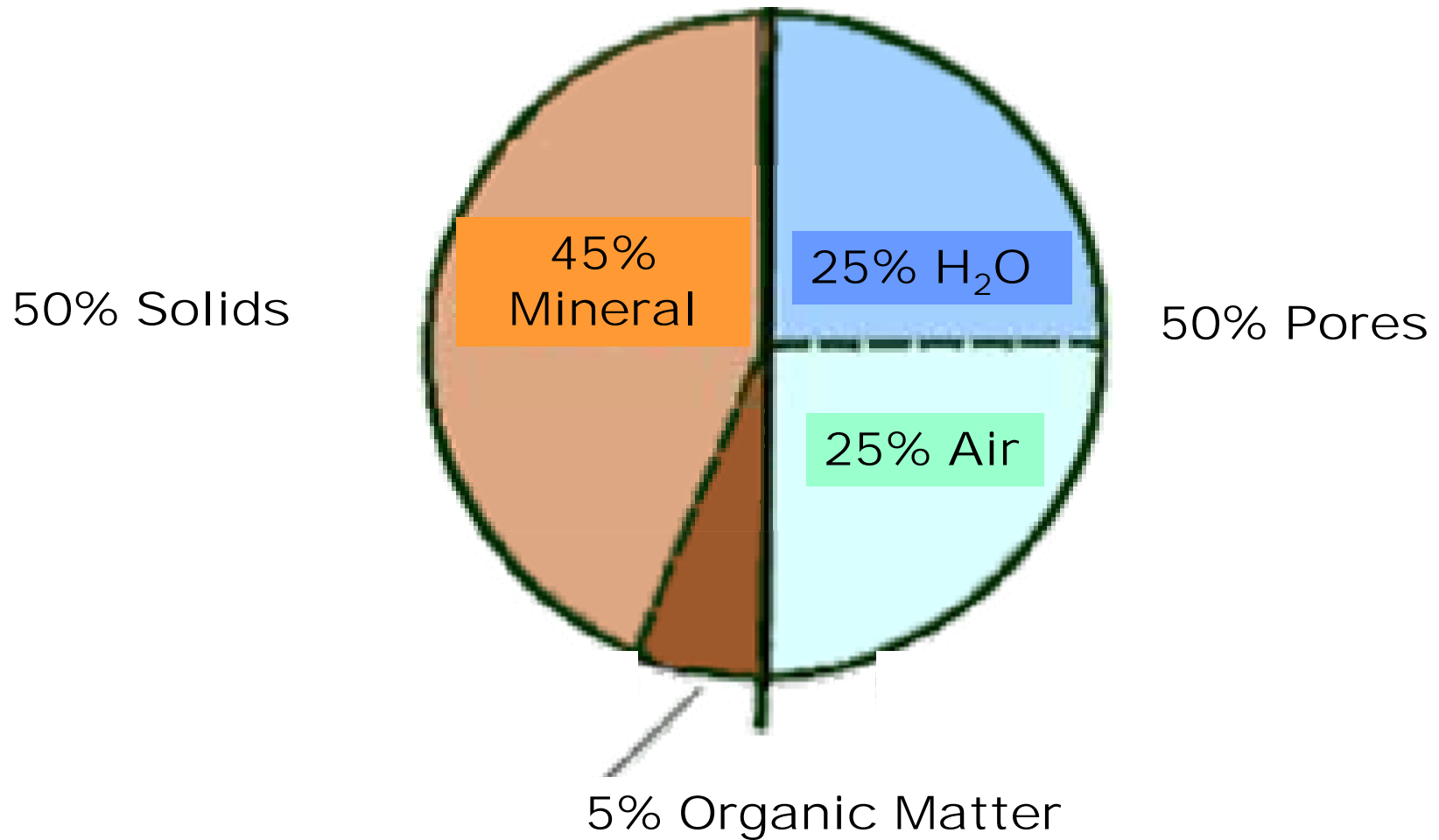
Micronutrients: B, Cu, Fe, Mn, Zn, Mo, Ni,
Co, Cl

Soil Factors Affecting Plant Nutrients

- Type and amount of clay
- pH
- Organic matter
- Water



Idealized Soil Composition



Composition of Hamakua Soils

- Defy conventional rules of soil science
- Why?
 - Made from weathered volcanic ash with exceptionally high surface area
 - High organic matter content (20%)
 - Very low bulk density (0.4)

Clay is Where the Action is!

Clay Properties:

Microscopic size (<0.002 mm)

Extremely high surface area

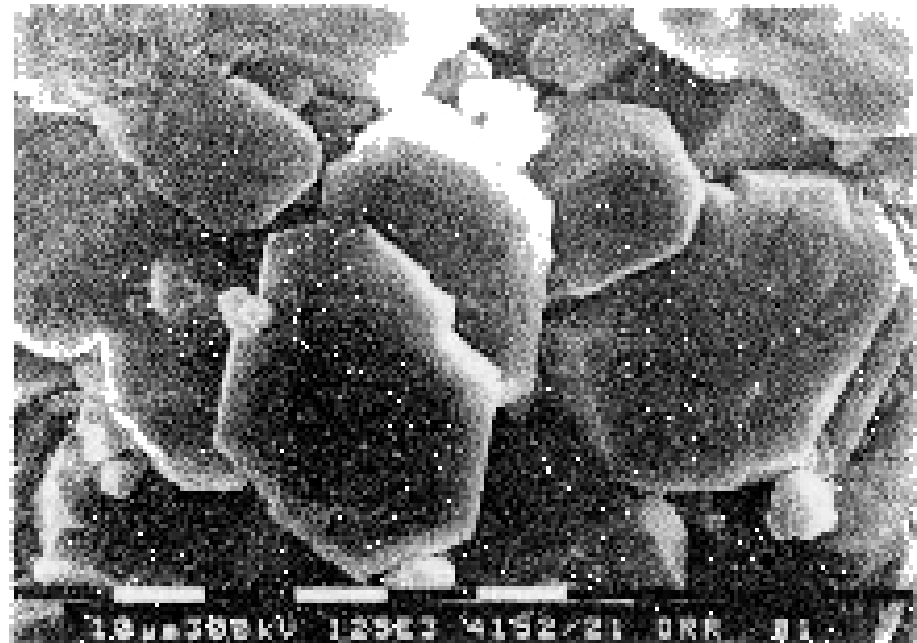
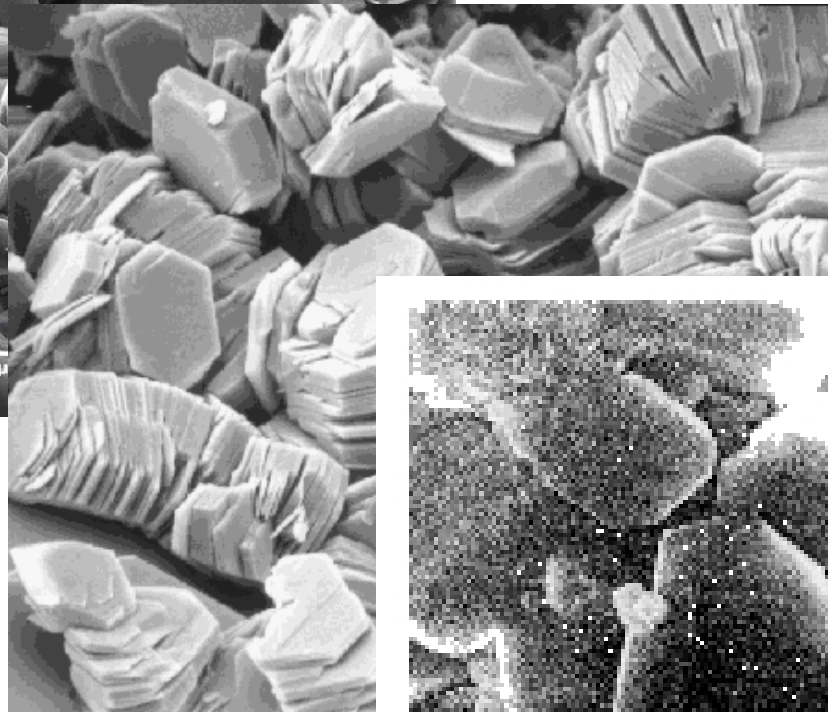
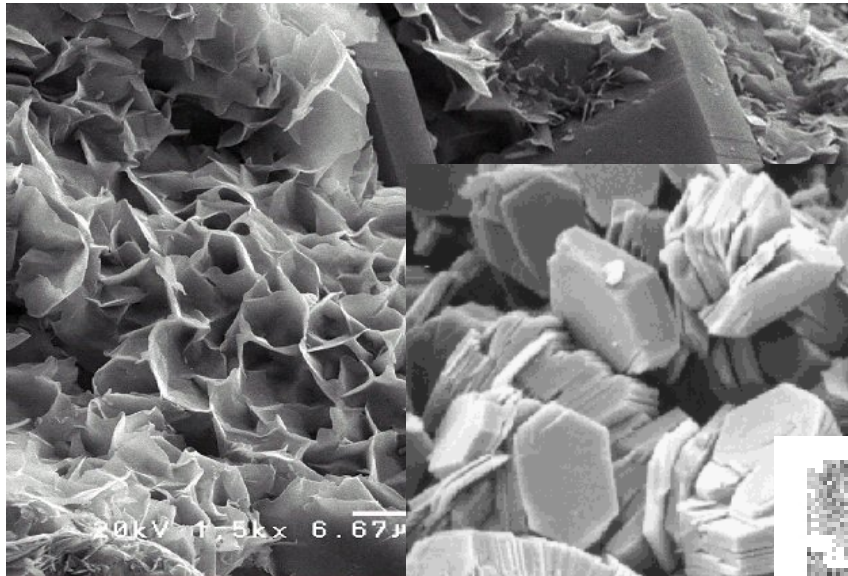
- water retention
- chemical reactions
- biological activity

Clay surfaces carry charge (-/+)

In Hawaii Type of Clay is Critical

2:1 layer silicates

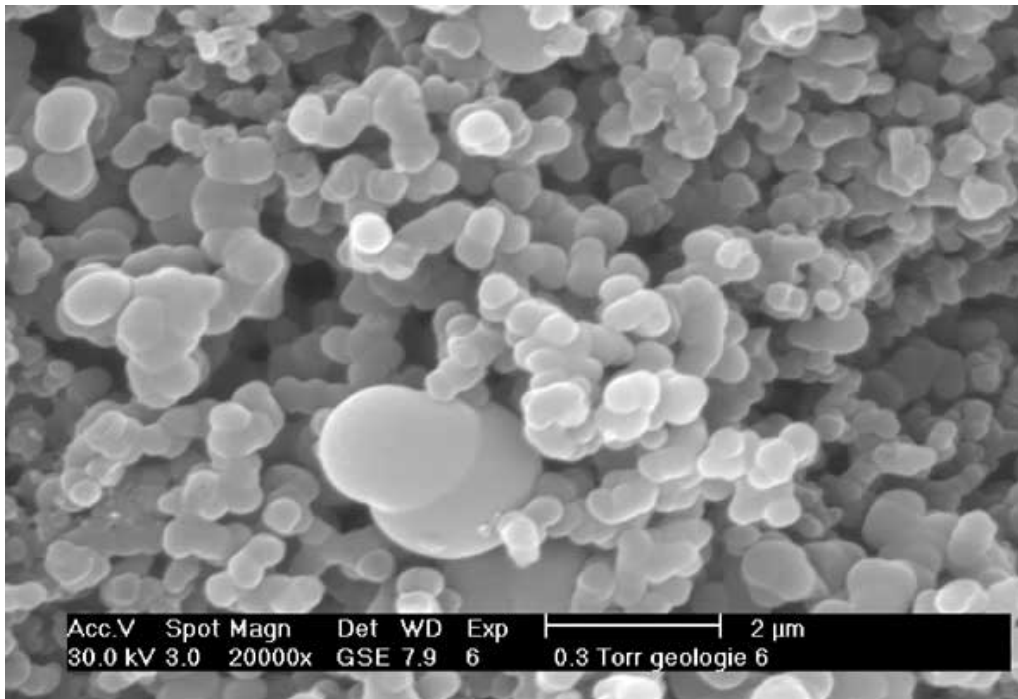
1:1 layer silicates



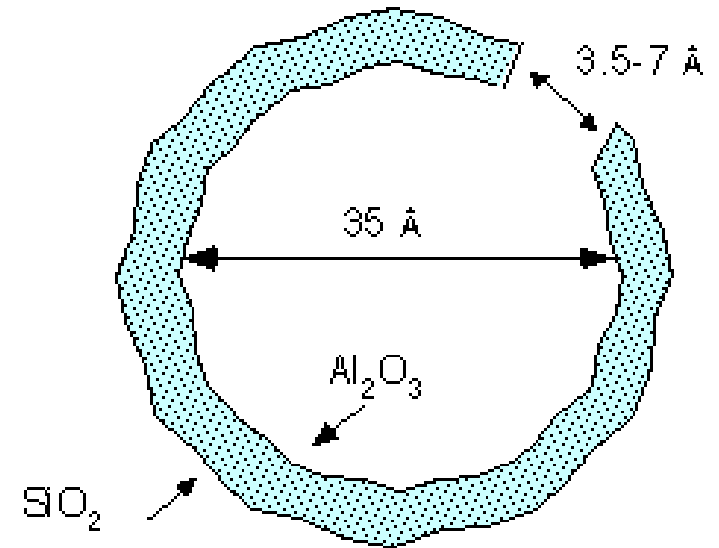
oxides

Volcanic Ash Soils Contain Allophane

Allophane



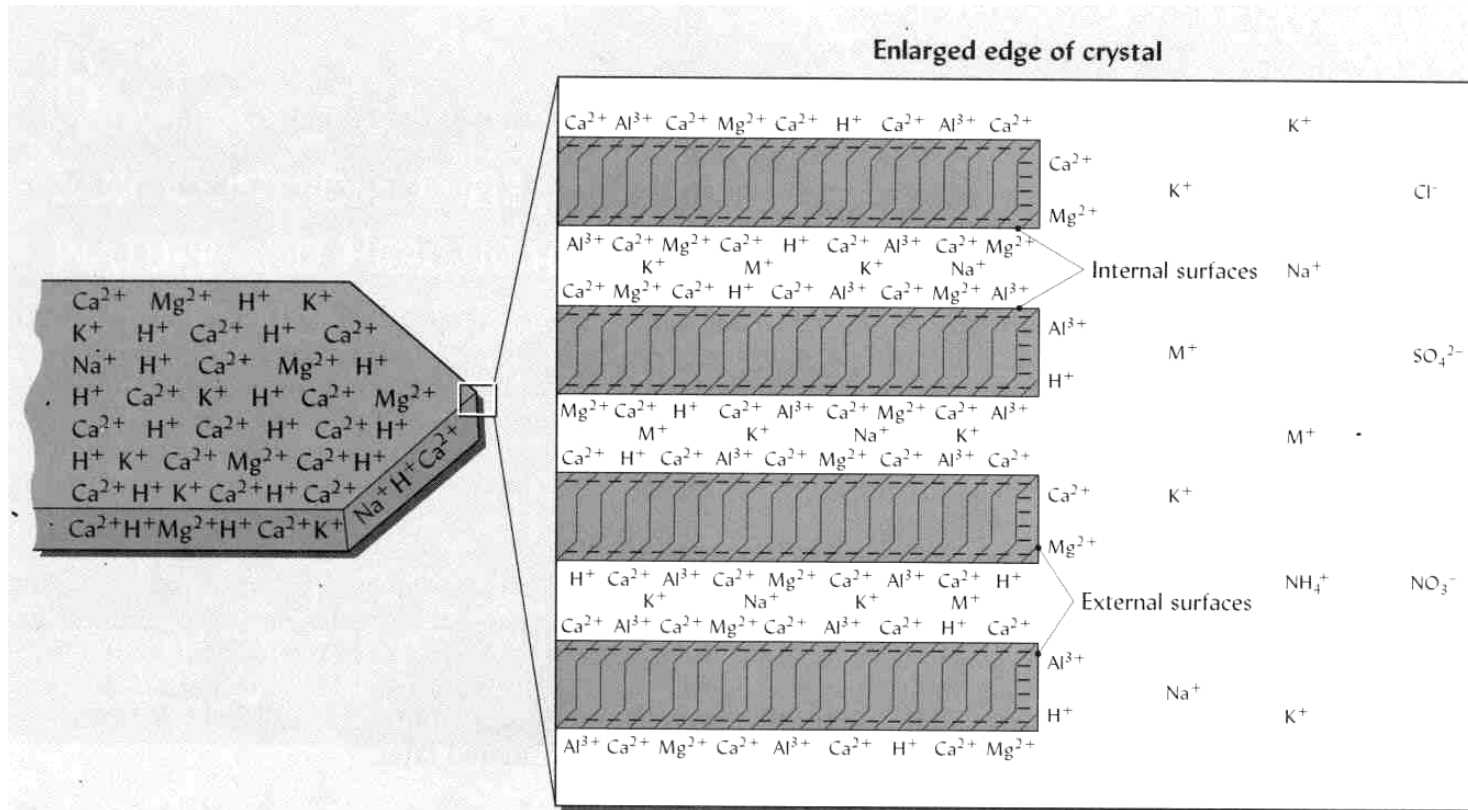
http://www.naturalsciences.be/institute/structure/geology/gsb_website/products/geolbelgica/publication/vol7a/goe



Properties:

- Exceptionally high surface area
- Variable charge (-/+)
- Low CEC

Cation Exchange Capacity CEC

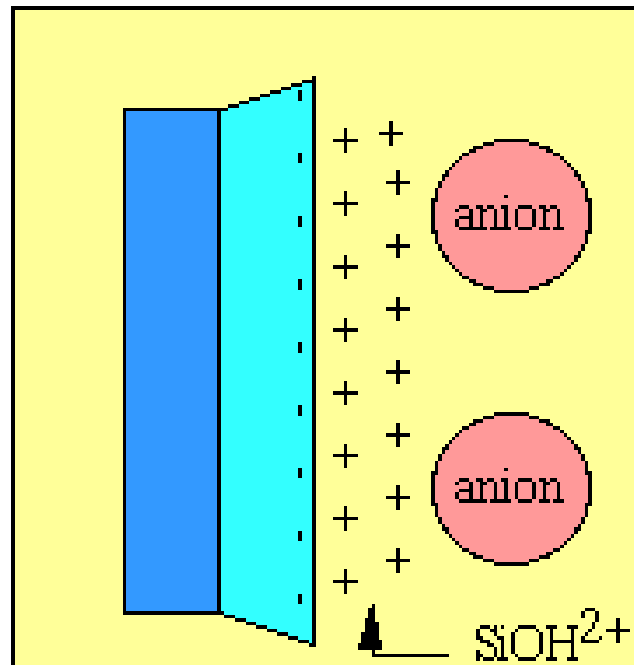


Brady & Weil, 2004. Elements of the Nature and Properties of Soils

Negatively charged sites that adsorb cations:
 Ca^{2+} , Mg^{2+} , K^+ , NH_4^+

Allophane and CEC

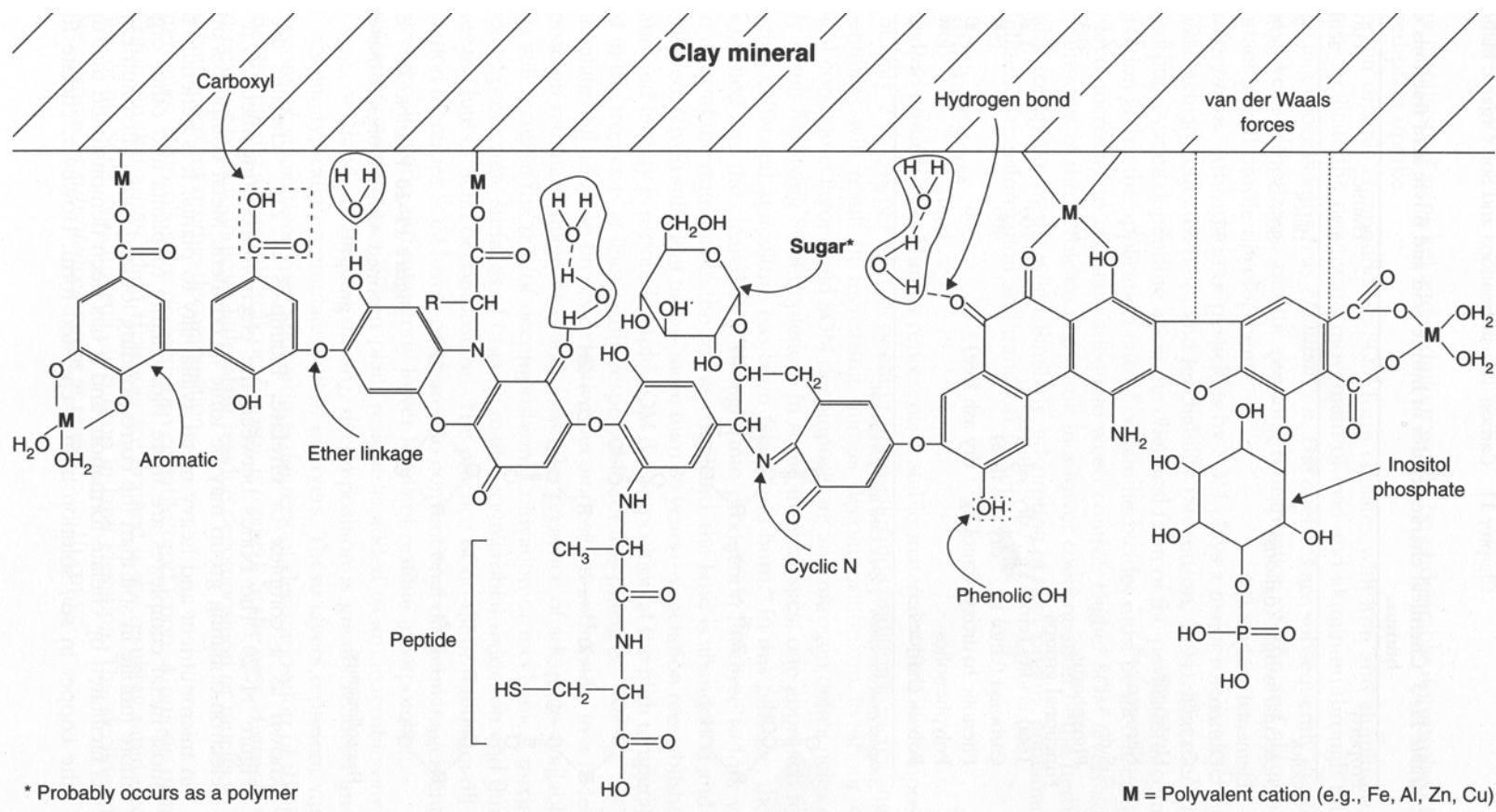
- When allophane is present surface charge can be - (CEC) or + (AEC)
 - Under acid conditions + charges increase
 - This results in a less fertile soil



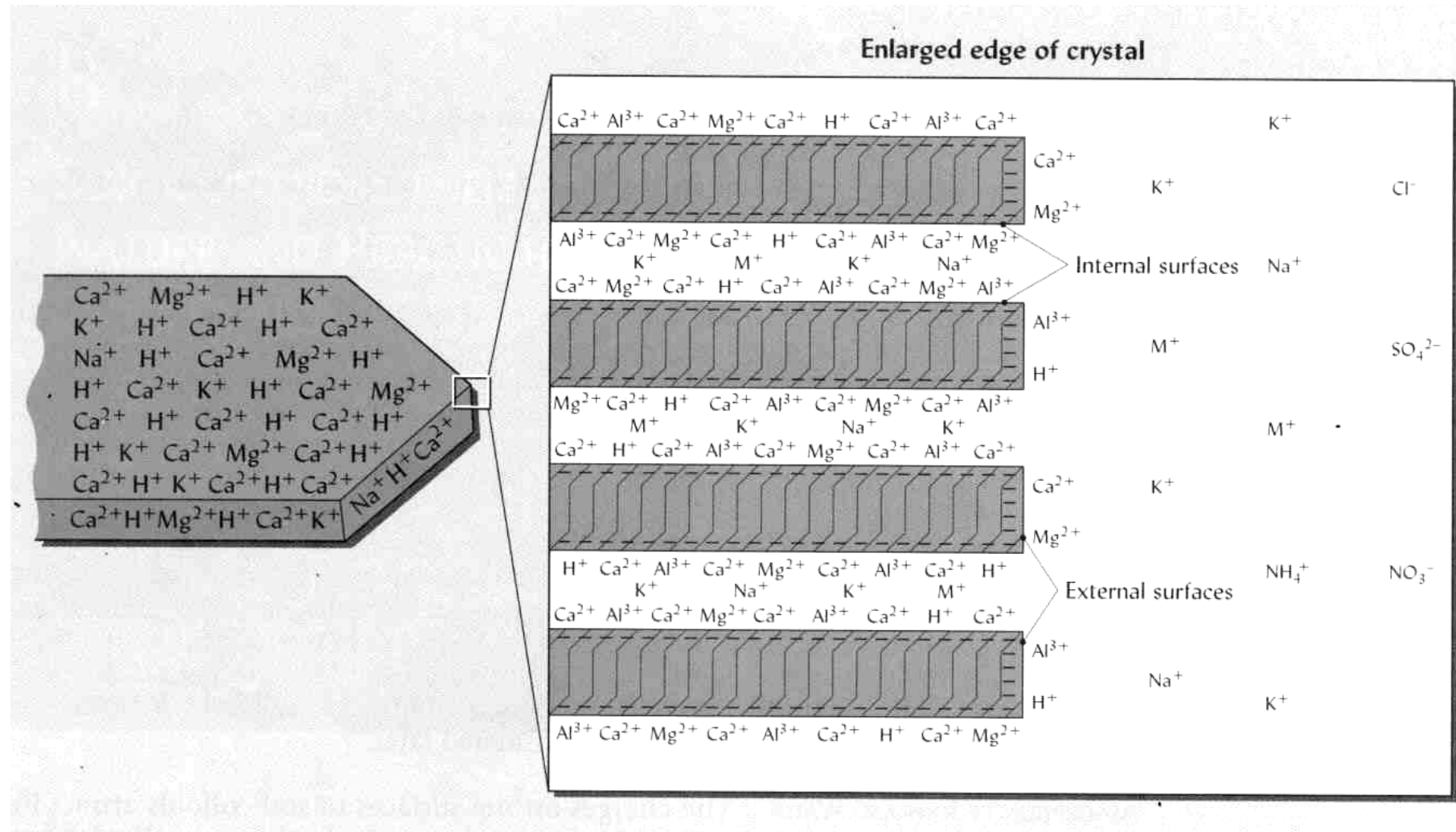
Organic Matter and CEC

- Organic matter has a high CEC
 - OM high in surface horizon
 - CEC always higher in surface soil

Brady & Weil, 2004. Elements of the Nature and Properties of Soils



CEC is a Bank for Nutrients



Soil Acidity

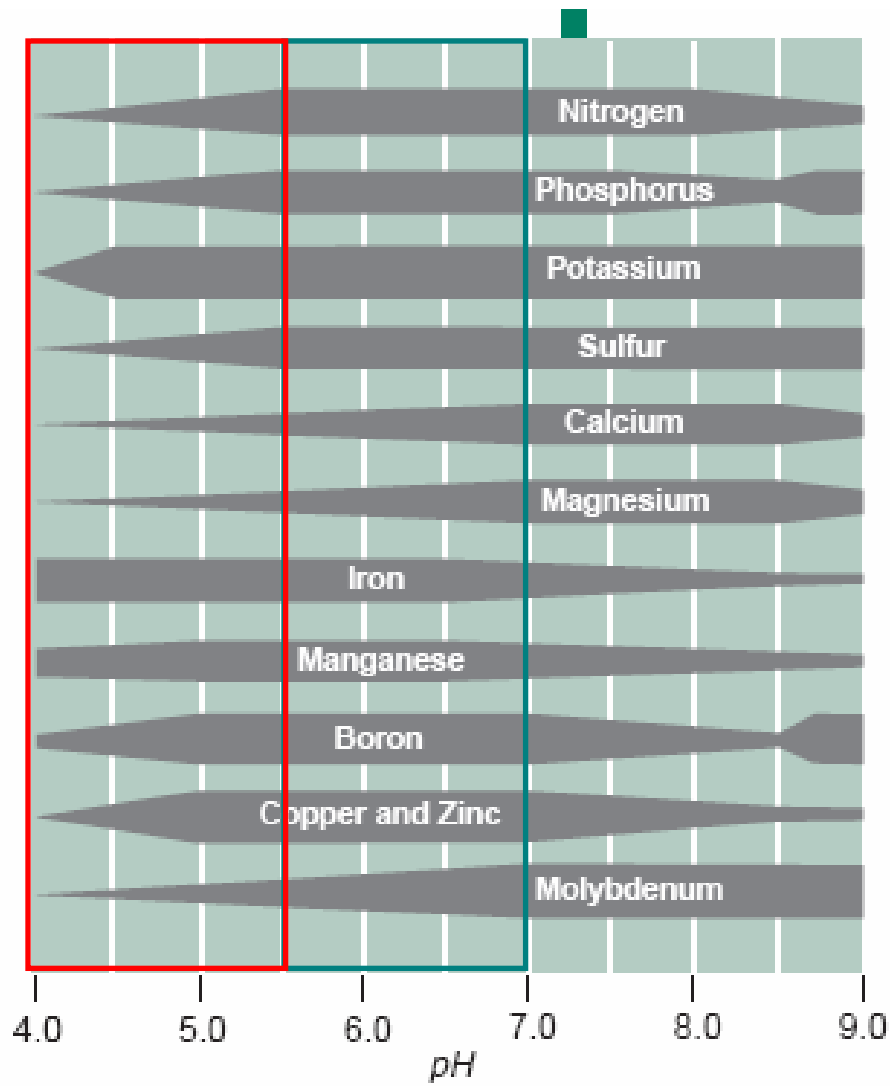
Natural Sources of Acidity:

- Precipitation and cation leaching
- Carbonic acid and organic acids
- Organic matter

Human Induced Acidity:

- Acid rain
- Urea
- Ammonium fertilizers
- Mono and diammonium phosphate
- Elemental S

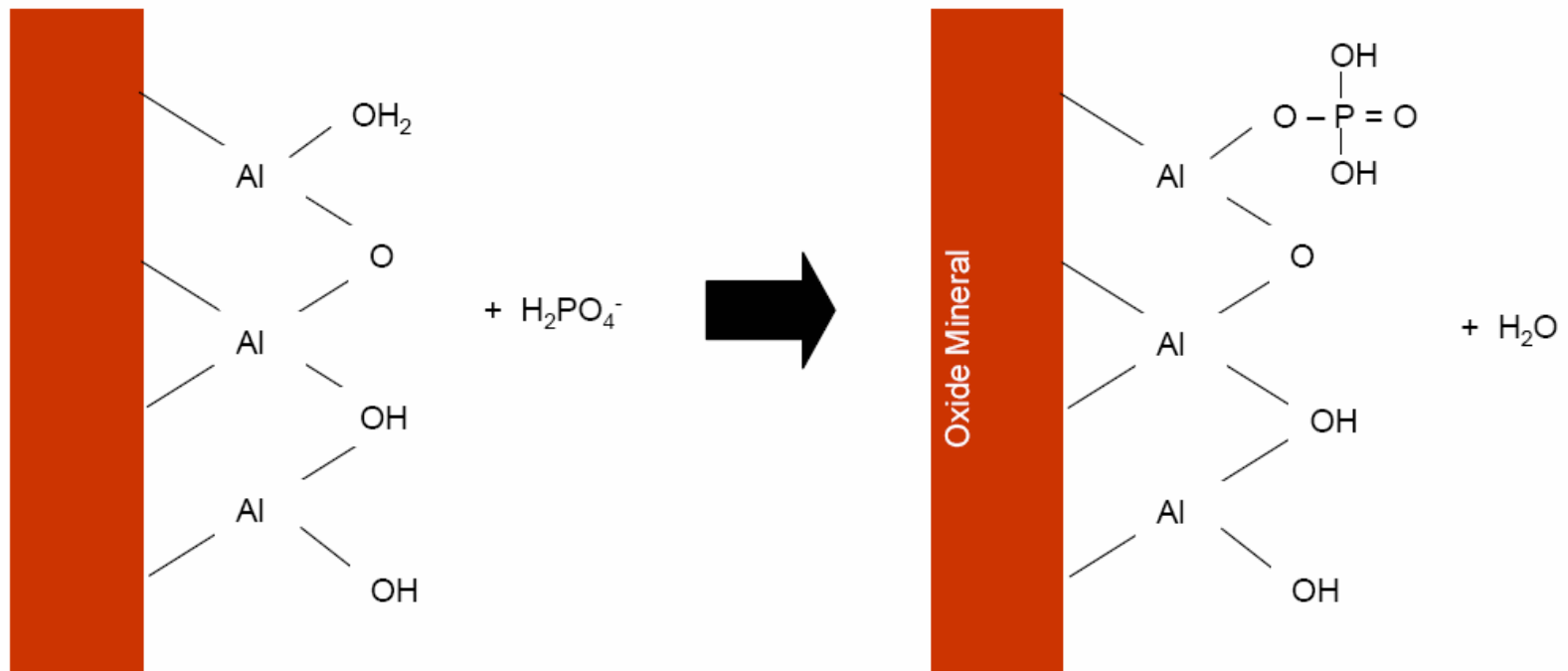
Soil pH and Nutrient Availability



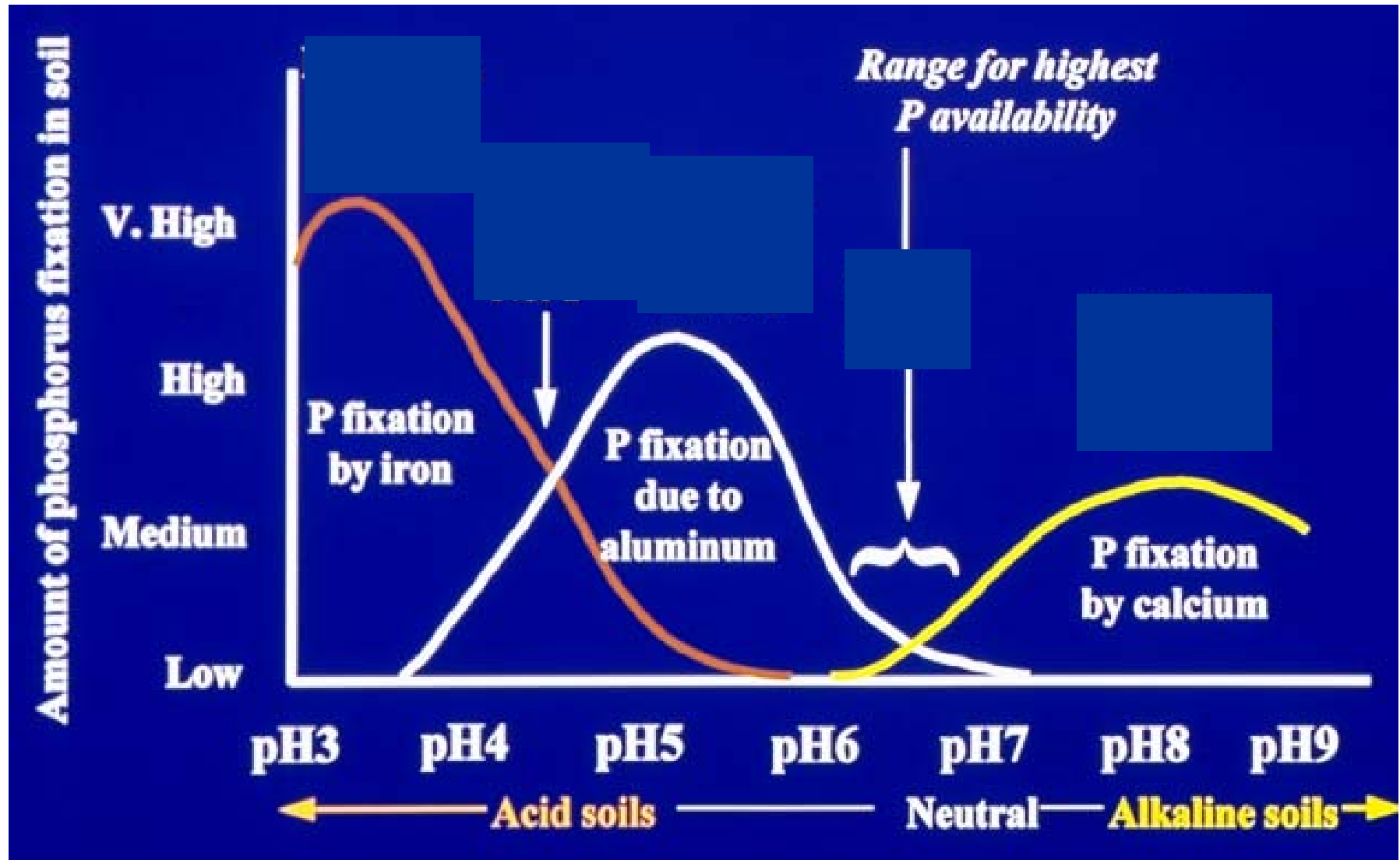
- Soil pH controls nutrient solubility
- Ideal range 6.0-6.5
- CEC decreases at low pH
- P fixation increases at low pH

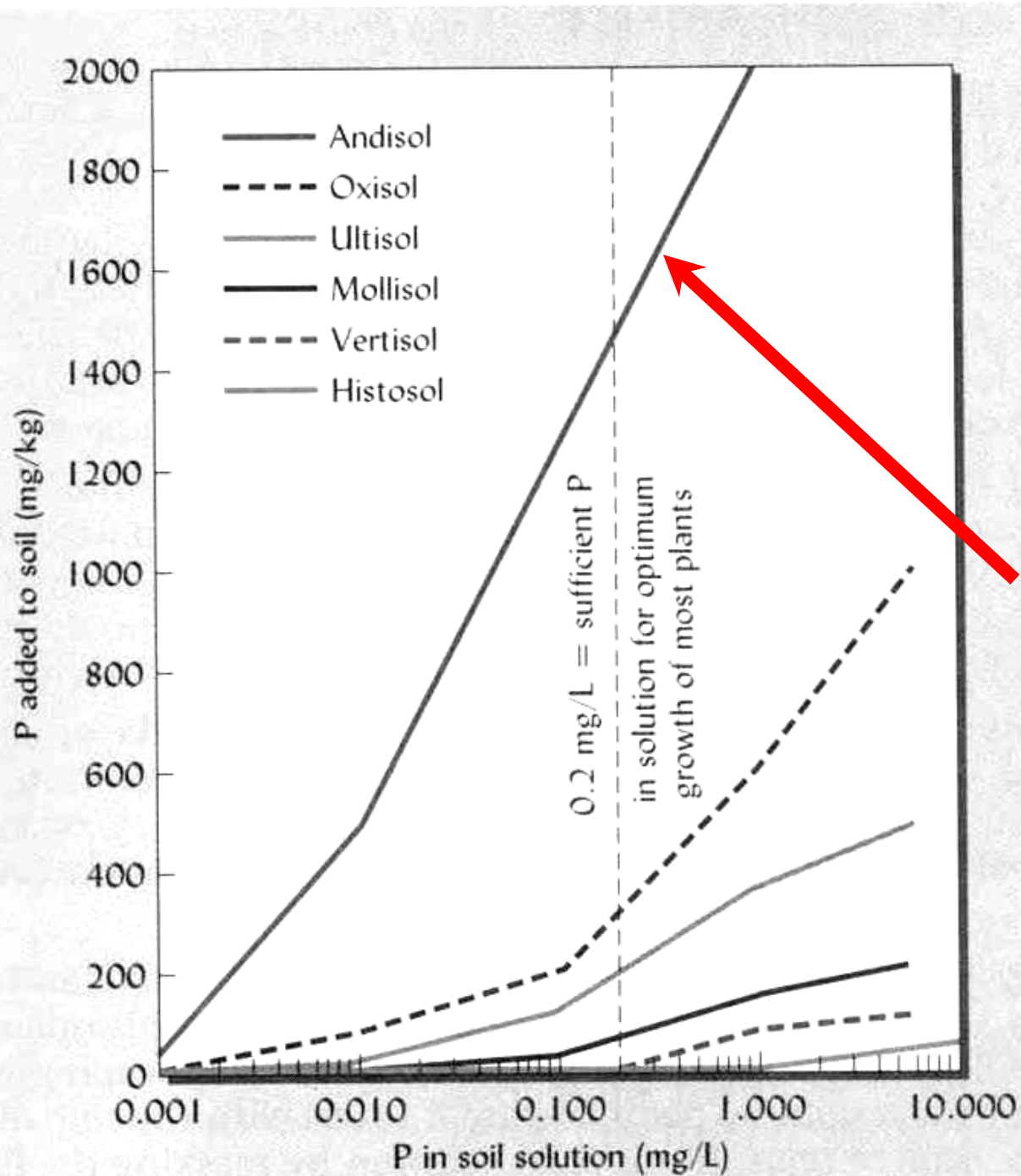
P-Fixation

- Adsorption of P on Oxides



P Fixation Depends on Soil pH



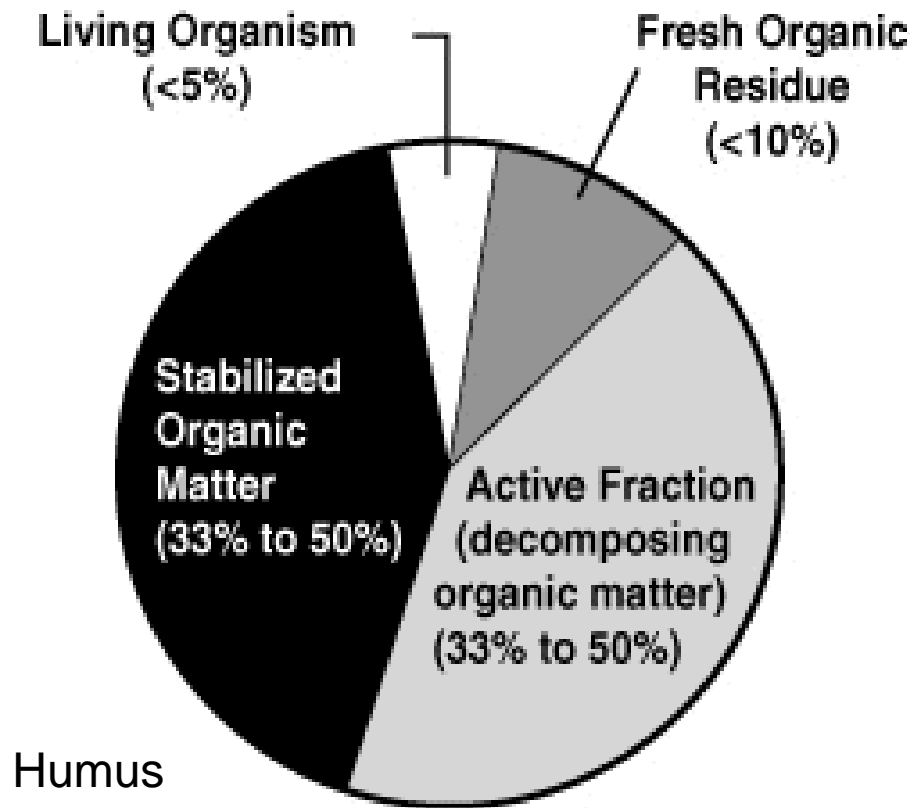


P Fixation is extremely high in ash soils

P-Fixation

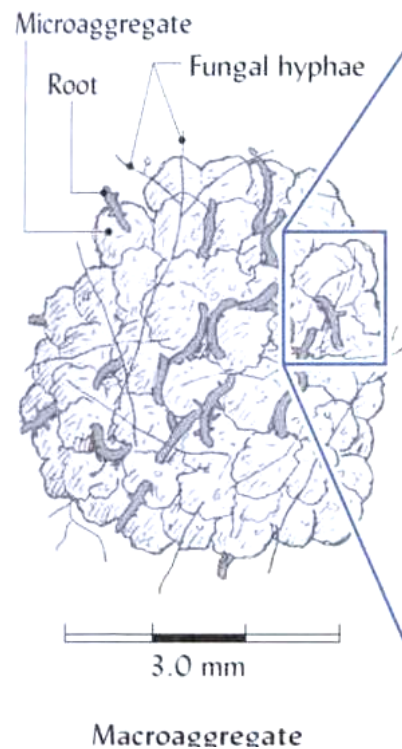
- Chemical bonding of phosphate with Aluminum/iron on clay surfaces
- Extremely high in wet volcanic ash soils (Hamakua)
- P-fixation increases as soils become more acid
- High P-fixing soils require high P inputs

Organic Matter



Importance of Organic Matter in Soils

- Physical Properties
 - Improves aggregation (glue)
 - Improves water holding capacity (surface area)



Importance of Organic Matter in Soils

- Chemical

- Increases nutrient availability (N cycling, P and micronutrient solubility)
- Increases CEC ($200 \text{ cmol}_c \text{ kg}^{-1}$)
- Buffers the soil against pH changes

N mineralization

Conversion of organic N to inorganic N

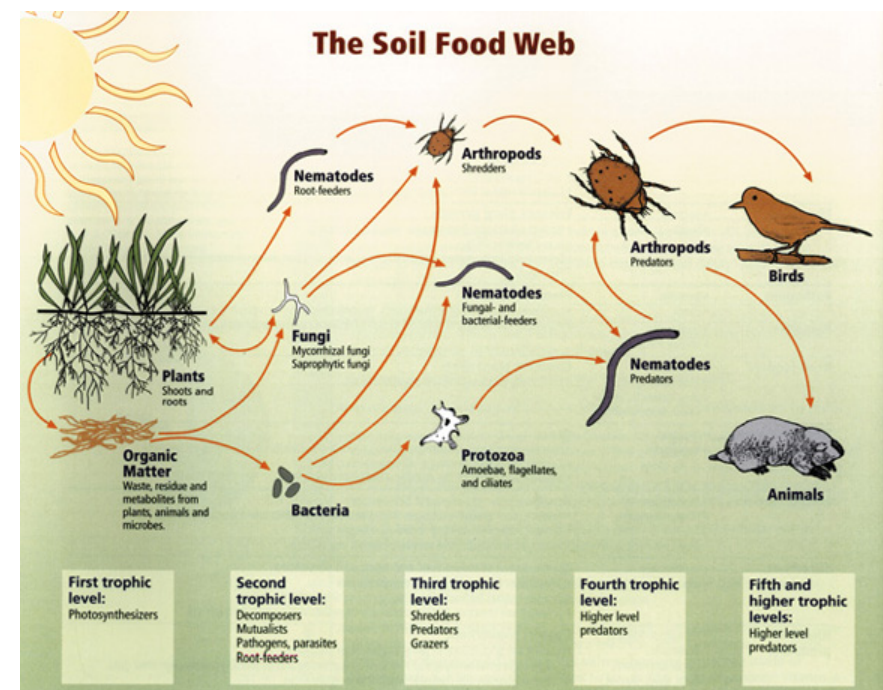


Total N mineralization

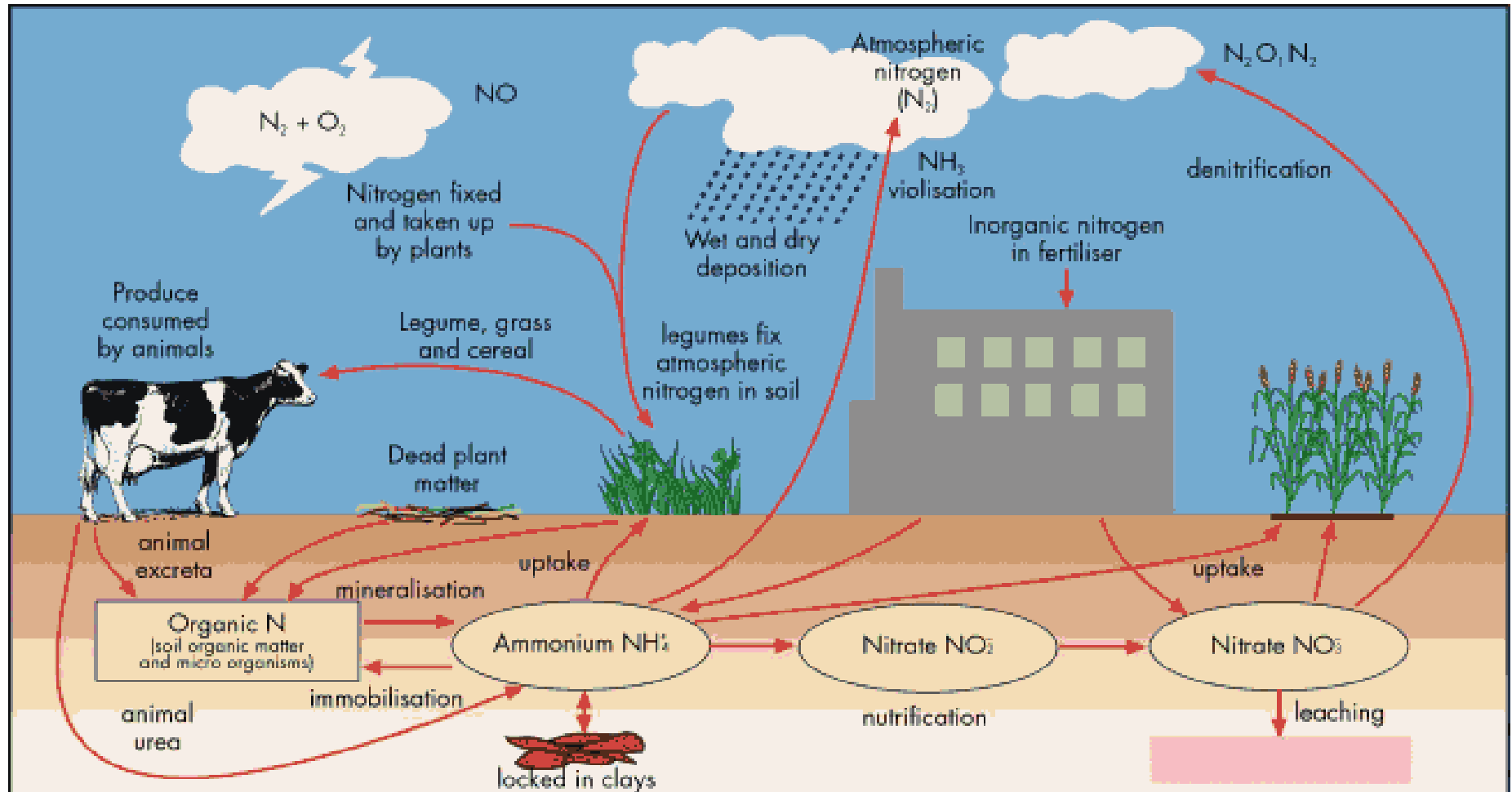
Importance of Organic Matter in Soils

- Biological

- N fixation
- Increases microbial diversity
- Assists in pathogen suppression



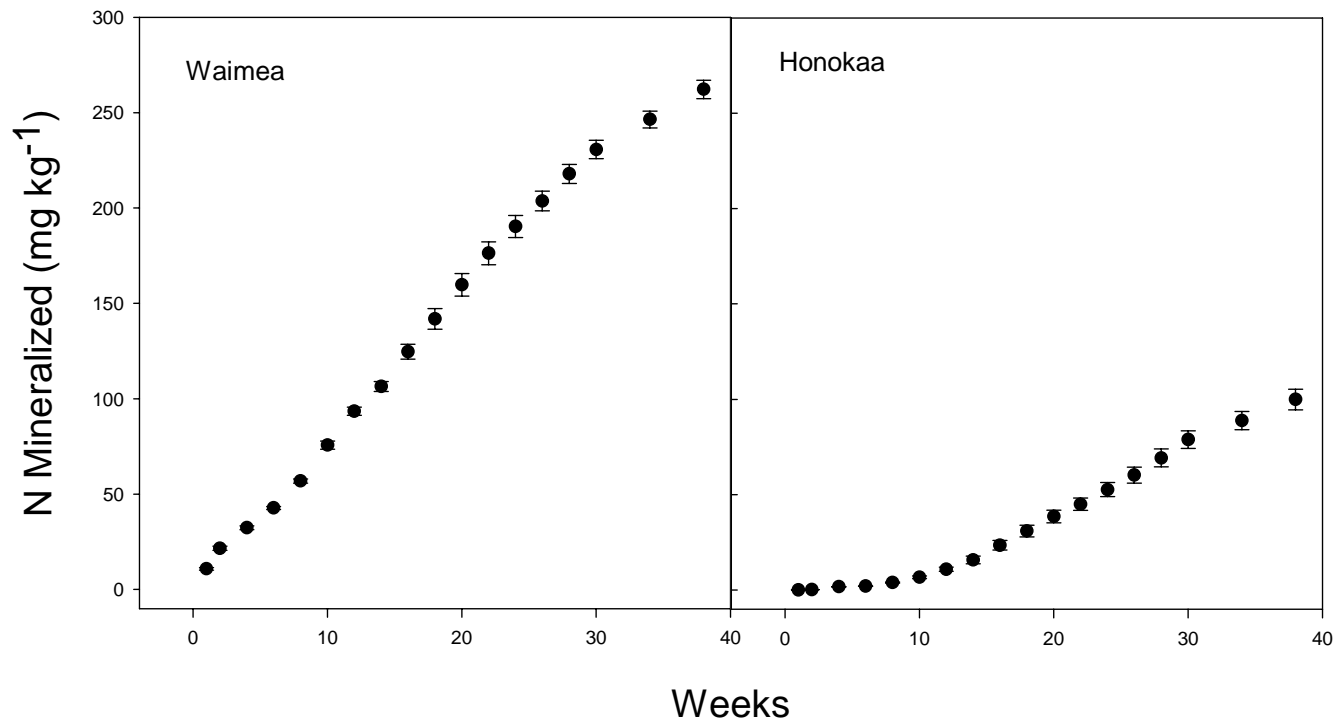
Nitrogen



Nitrogen Dynamics



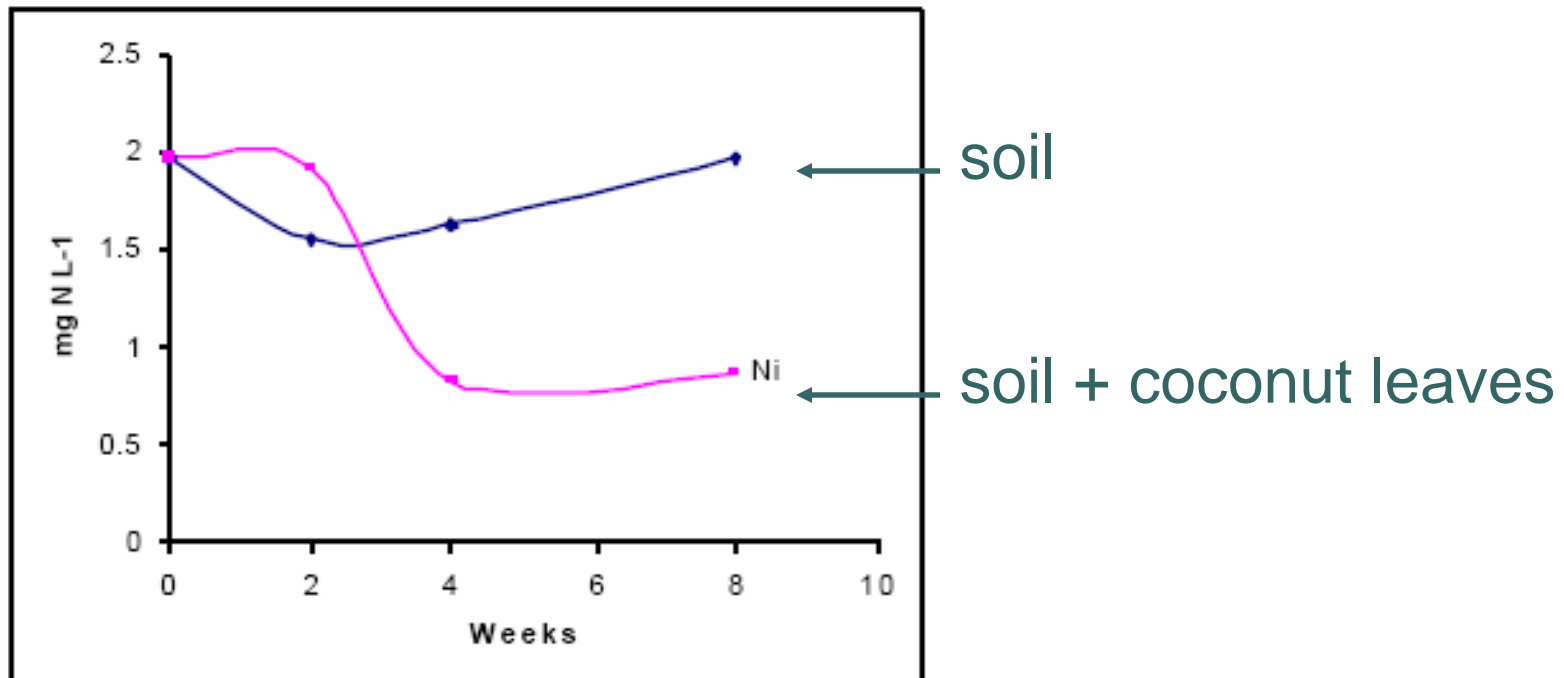
Mineralization: microbial conversion of organic N into plant available inorganic forms (NH_4^+ , NO_3^-)



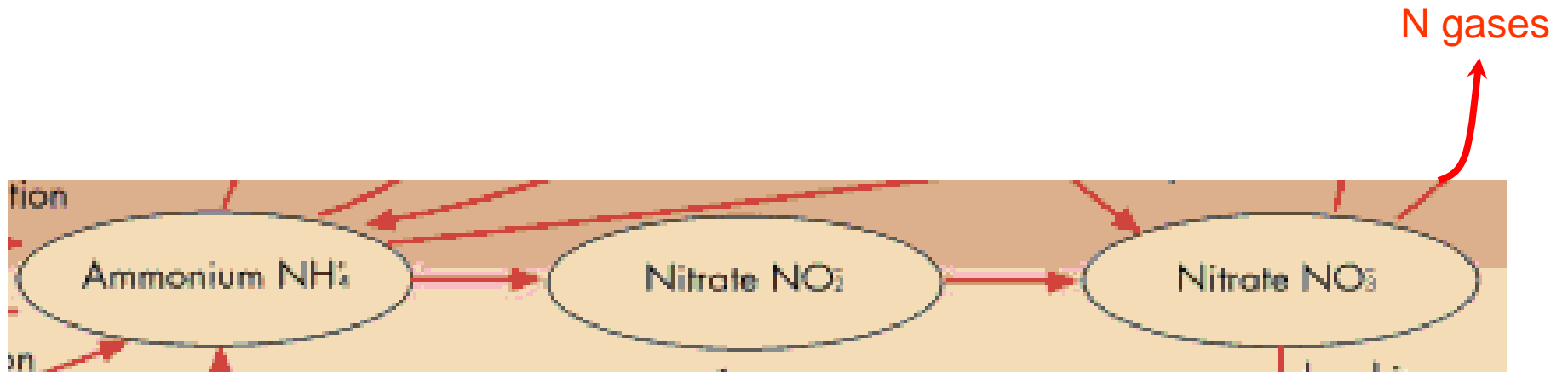
Nitrogen Dynamics



Immobilization: Conversion of inorganic N to organic N by microbes



Denitrification: Conversion of NO_3^- into N_xO gases

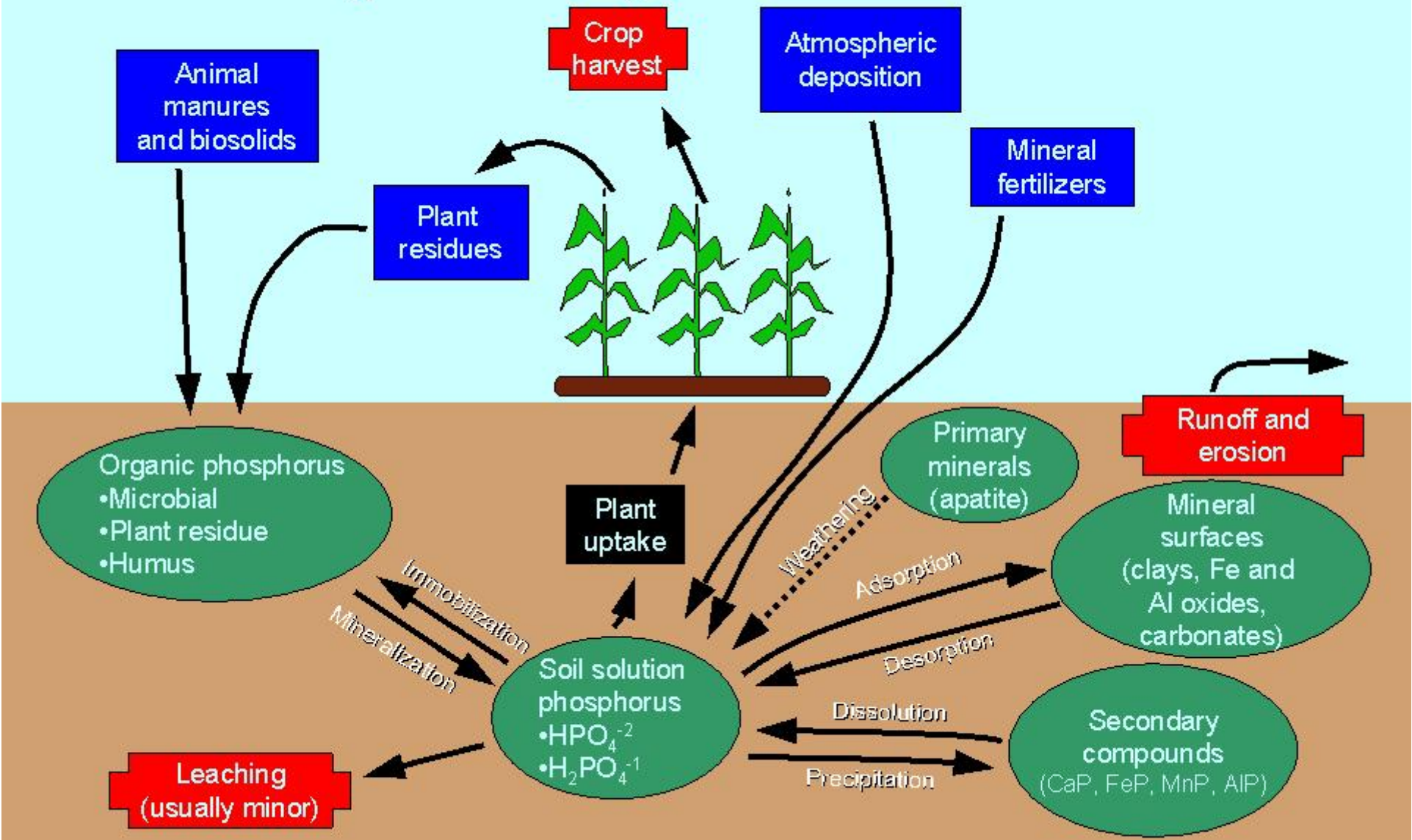


Denitrification occurs when:

- Soils are saturated and lack oxygen
- Denitrification rate increases with temperature and organic matter content

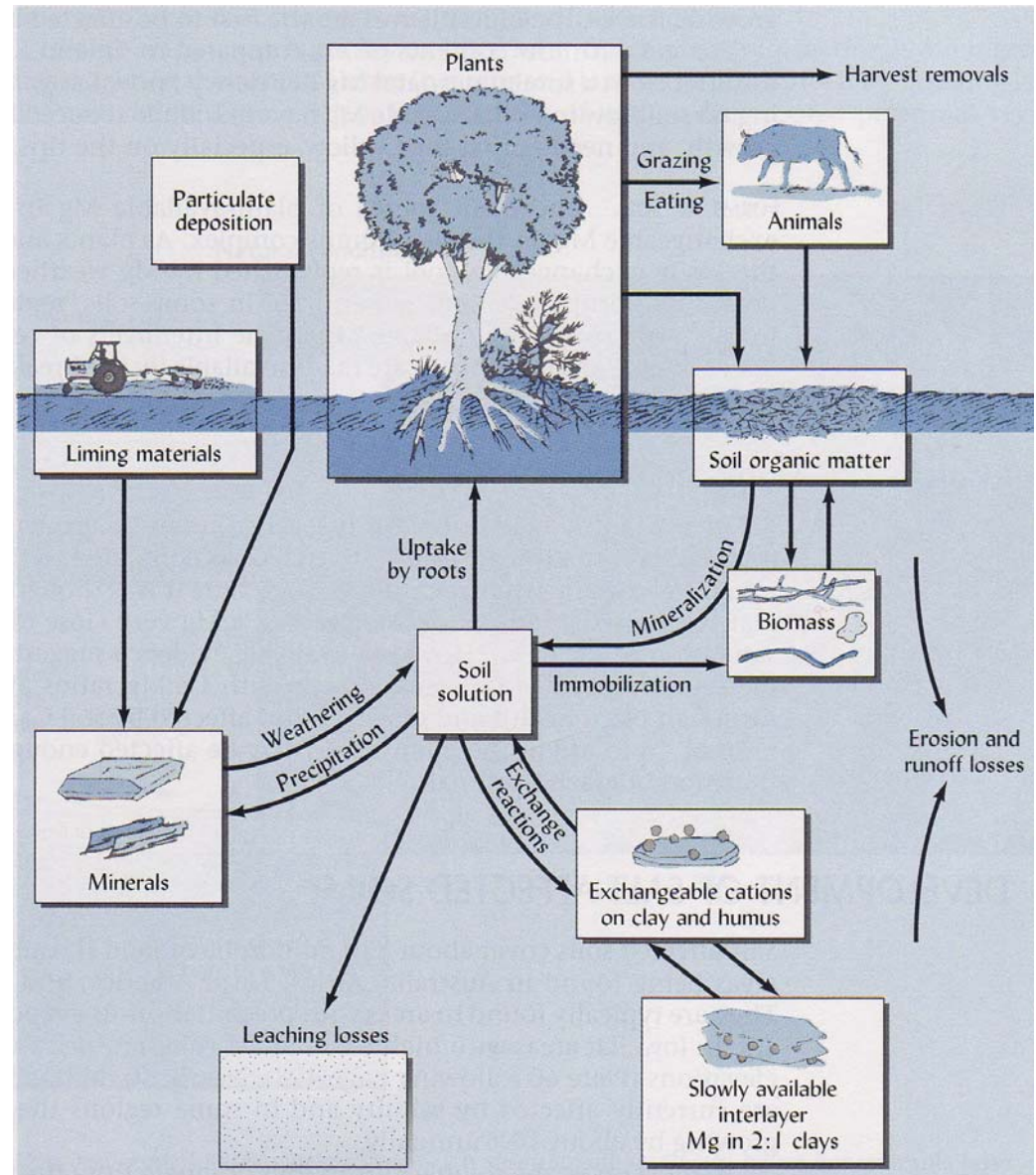
The Phosphorus Cycle

Component Input to soil Loss from soil



Calcium, Magnesium & Potassium

- Sources:
 - Minerals
 - Organic matter
- Availability:
 - CEC
 - pH
 - Leaching potential



Fertile Soils

- Slightly acid to neutral pH
- High CEC
- High organic matter

Vertisols - *Lualualei*

Mollisols - *Waialua*

Medial Andisols -
Waimea

Infertile Soils

- Acid to strongly acid pH
- Low CEC
- High P fixation

Soil Fertility Depends on:

Fertile Soils

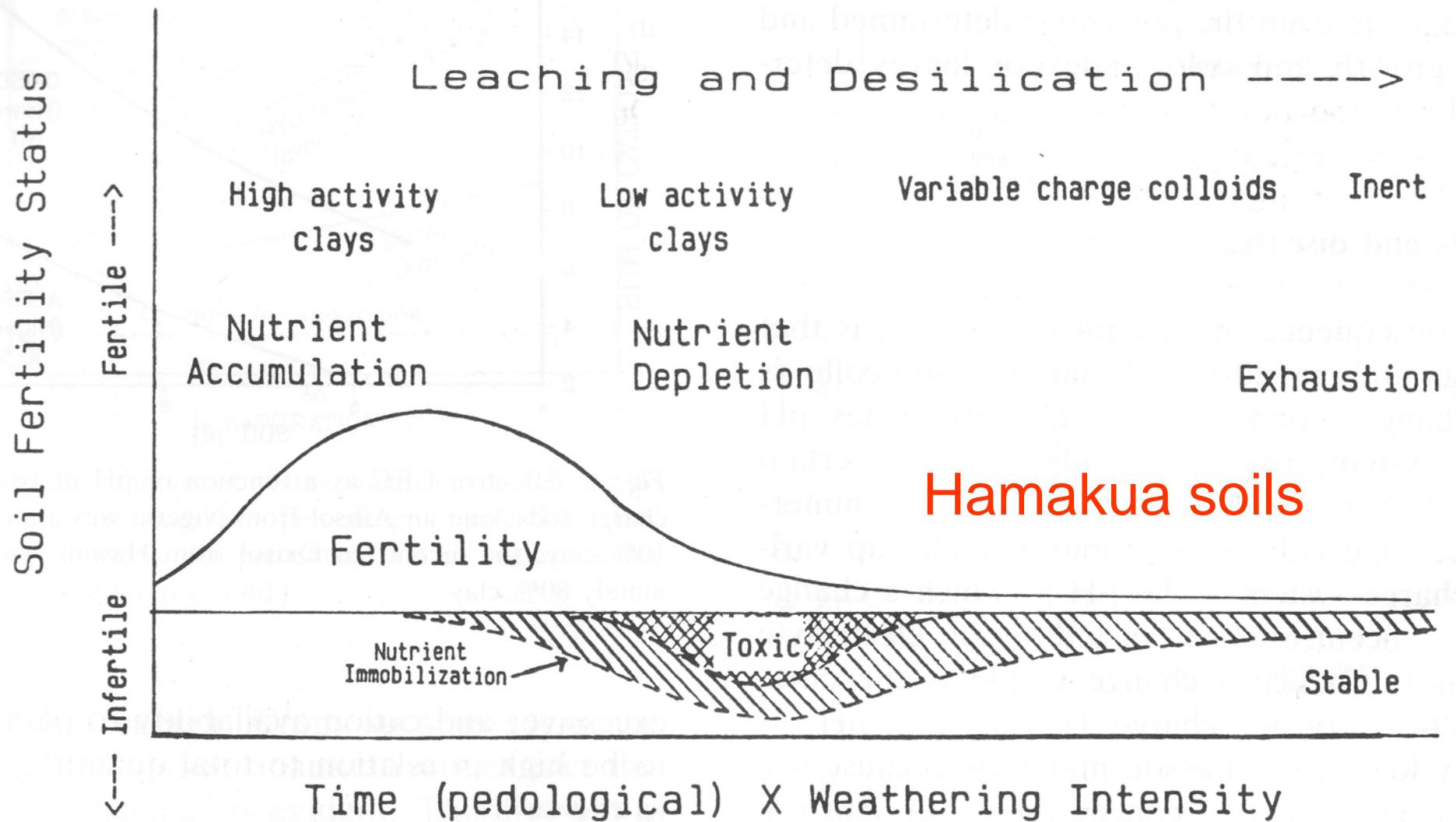
- Slightly acid to neutral pH
- High CEC
- High organic matter

Infertile Soils

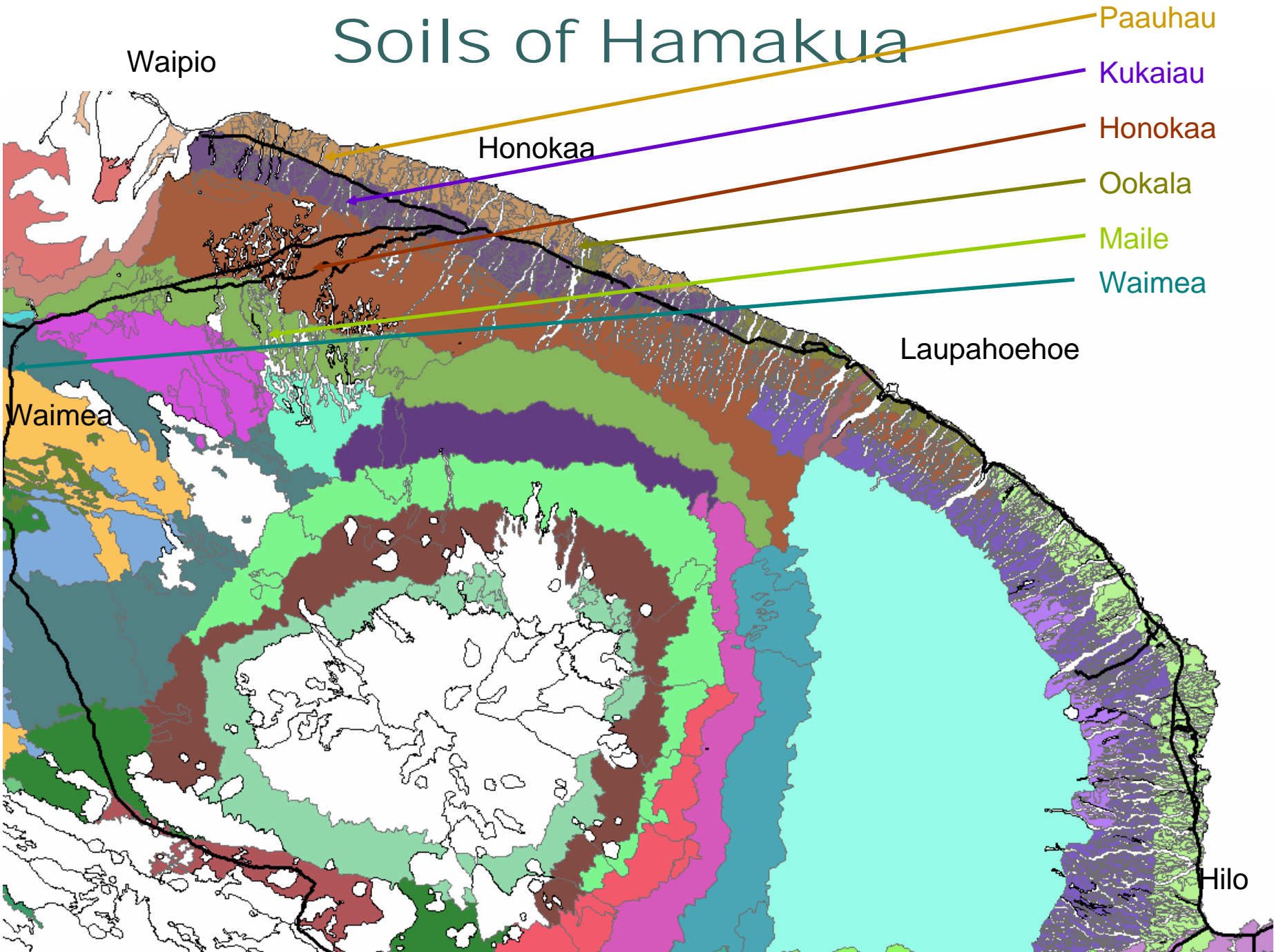
- Acid to strongly acid pH
 - Low CEC
 - High P fixation
-
- Amount of clay
 - Type of Clay
 - Organic Matter

Weathering Intensity and Fertility

TROPICAL SOIL FORMATION AND DEGRADATION

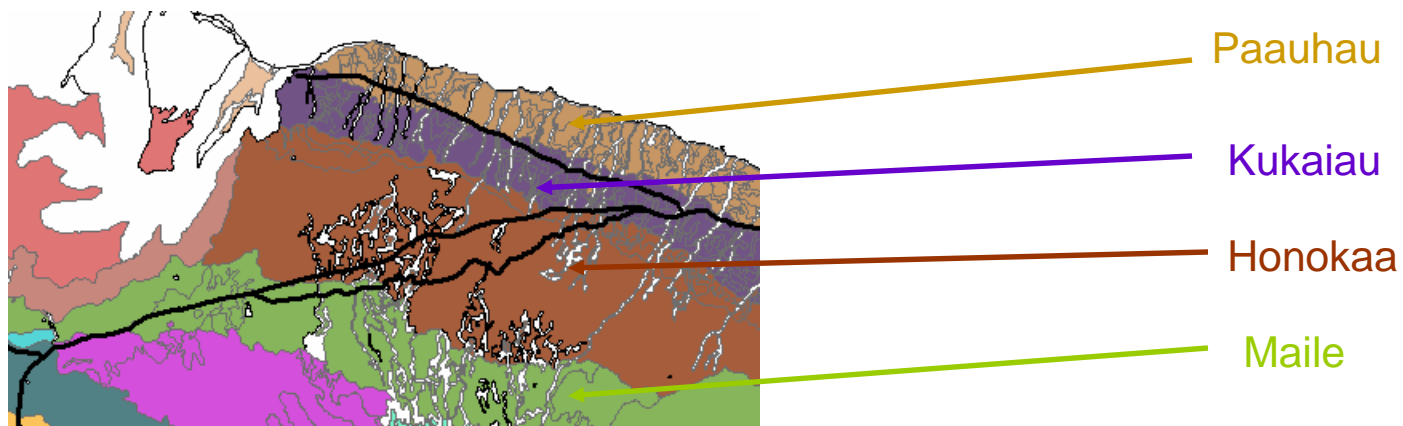


Soils of Hamakua



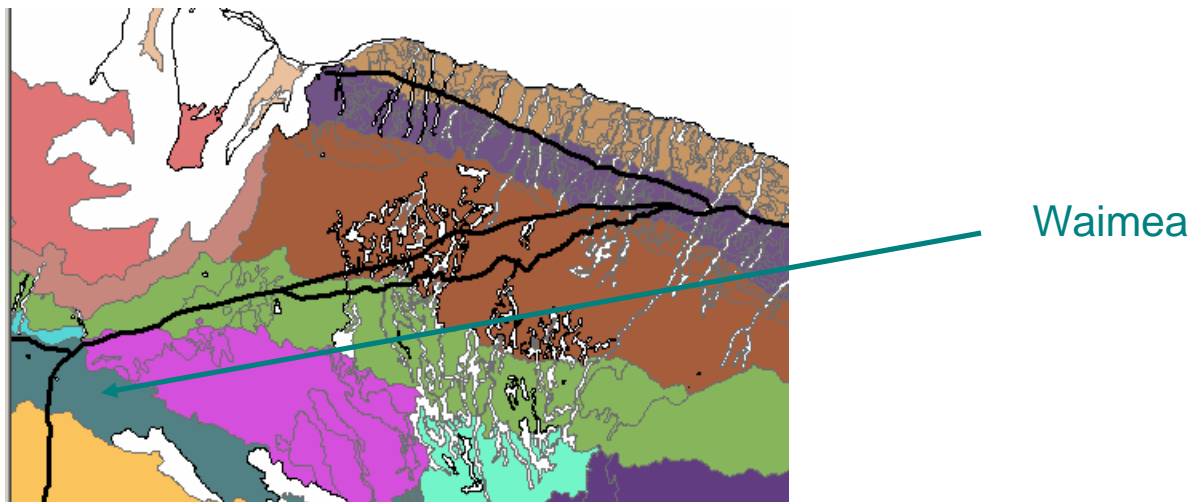
Fertility Status

Soil	Rainfall in	pH	Organic C %	Total N	Ca ppm
Paauhau	60-80	5.8	5.4	0.6	2,960
Honokaa	120-150	5.4	11.7	0.9	1,880
Maile	60-80	5.7	15.5	1.1	520



Fertility Status

Soil	Rainfall	pH	Organic C	Total N	Ca
	in		%		ppm
Waimea	30	6.4	13.1	1.2	6,000



Summary

- Soils give us life
- Type of clay determines fertility
- Organic matter makes a difference
- Some soils are inherently more fertile than others
- If we know our soils we can manage them well